

1-(4-Methylphenyl)-3-(1H-1,2,4-triazol-1-yl)-
propan-1-oneJun Wan, Chun-Li Li, Xue-Mei Li,
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In the title compound, $C_{12}H_{13}N_3O$, the benzene and triazole rings make a dihedral angle of $59.6(2)^\circ$. The crystal packing is stabilized by weak $C-H \cdots \pi$ interactions and van der Waals forces.

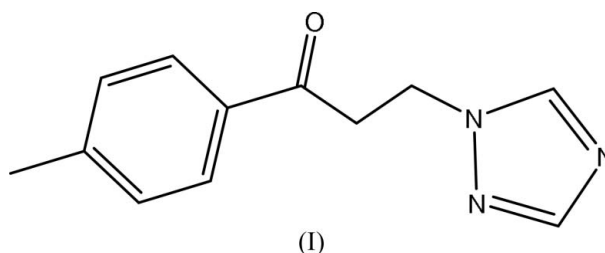
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Comment

In our ongoing studies of triazole compounds, the title compound, (I), was obtained in the reaction of triazole and 3-(dimethylamino)-1-(4-methylphenyl)propan-1-one hydrochloride. We report its crystal structure here.



Key indicators

Single-crystal X-ray study

 $T = 293$ KMean $\sigma(C-C) = 0.003$ Å R factor = 0.037 wR factor = 0.108

Data-to-parameter ratio = 8.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The bond lengths and angles in (I) (Table 1) are within normal ranges (Allen *et al.*, 1987) and are comparable with those in related compounds (Wan, Li, Li & Zhang, 2005; Wan, Li, Li, Li *et al.*, 2005). The molecular skeleton of (I) is non-planar, with a dihedral angle of $59.6(2)^\circ$ between the benzene and triazole rings.

There is a weak intramolecular $C9-H9A \cdots N2$ hydrogen bond (Table 2). The crystal packing is stabilized by weak intermolecular $C-H \cdots \pi$ interactions (Table 2) and van der Waals forces.

Experimental

Triazole (3.5 g, 0.58 mol) was added to a solution of 3-(dimethylamino)-1-(4-methylphenyl)propan-1-one hydrochloride (11.4 g, 0.05 mol) in water (30 ml). The mixture was heated under reflux for 4 h, yielding a copious precipitate. Colourless single crystals of (I) suitable for X-ray diffraction study were obtained by slow evaporation of an ethyl acetate–petroleum ether (1:1, v/v) solution over a period of two weeks.

Crystal data

 $C_{12}H_{13}N_3O$ $M_r = 215.25$ Orthorhombic, $Pca2_1$ $a = 10.943(4)$ Å $b = 13.777(5)$ Å $c = 7.514(3)$ Å $V = 1132.8(7)$ Å³ $Z = 4$ $D_x = 1.262$ Mg m⁻³Mo $K\alpha$ radiationCell parameters from 2634
reflections $\theta = 2.4\text{--}25.2^\circ$ $\mu = 0.08$ mm⁻¹ $T = 293(2)$ K

Block, colourless

 $0.45 \times 0.26 \times 0.12$ mm

Data collection

Siemens SMART 1000 CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.963$, $T_{\max} = 0.990$
 5789 measured reflections

1220 independent reflections
 1067 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$
 $\theta_{\text{max}} = 26.2^\circ$
 $h = -12 \rightarrow 13$
 $k = -13 \rightarrow 16$
 $l = -9 \rightarrow 8$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.108$
 $S = 1.14$
 1220 reflections
 146 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0611P)^2 + 0.0695P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.13 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.13 \text{ e } \text{\AA}^{-3}$
 Extinction correction: SHELXTL (Sheldrick, 1997)
 Extinction coefficient: 0.024 (4)

Table 1

Selected bond lengths (Å).

O1—C8	1.215 (3)	N2—C11	1.311 (4)
N1—C12	1.323 (4)	N3—C12	1.321 (4)
N1—N2	1.350 (3)	N3—C11	1.325 (5)
N1—C10	1.460 (3)		

Table 2

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the triazole ring.

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C9—H9A \cdots N2	0.97	2.63	2.988 (3)	102
C12—H12A \cdots Cg1 ⁱ	0.93	2.67	3.571 (3)	135

Symmetry codes: (i) $-x, -y + 1, z - \frac{1}{2}$.

All H atoms were located in a difference Fourier map and constrained to ride on their parent atoms, with C—H distances in the range 0.93–0.97 Å and with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$. Friedel pairs were merged before the final refinement.

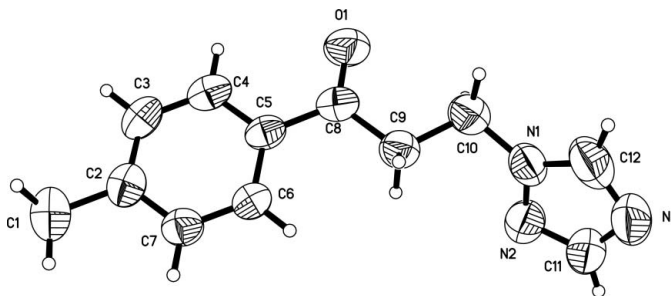


Figure 1

A view of (I), showing the atom-numbering scheme and 50% probability displacement ellipsoids.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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